

Rae Victor, Raphael Ben-Shoshan, and Shalom Sarel*: Photoinduced Formation of Vinylcyclohexatriene-Iron Carbonyl Complexes from Substituted Vinylbenzenes. Localization of Electrons in Aromatic Substrates *via* π Coordination to Metal.

Page 1934. Footnote *b*, legend of Table III. The footnote should read "*J* taken for all compounds (Hz): $H_{2,a}H_{2,b} = 2.2-2.7$; $H_{2,a}H_{1'} = 7.8-9.0$; $H_{2,b}H_{1'} = 6.7-7.2$; $H_2H_3 = 5.0-5.7$; $H_2H_6 = 1.5-1.9$; $H_3H_4 = 6.0-6.3$; $H_3H_5 = 1.5-2.7$; $H_4H_5 = 4.5$; $H_5H_6 = 6.0-6.9$."

W. Herz,* S. V. Bhat, and A. Srinivasan: Berlandin and Subacaulin, Two New Guianolides from *Berlandia subacaulis*.

Page 2532. Abstract. The last sentence of the abstract should read "Berlandin is either **1b** or differs from acetylsubacaulin (**2b**) in configuration of the epoxide ring."

H. W. Heine,* P. G. Williard, and T. R. Hoye: The Synthesis and Reactions of Some 1-(Nitroaryl)diaziridines.

Page 2982. Column 2. In line 25, "**22**" should read 2-isopropyl-6-nitrobenzotriazole 1-oxide; in lines 34 and 35, "**23**" should read 2-cyclohexyl-6-nitrobenzotriazole 1-oxide; in line 58, "**24**" should read 1-(2,4-dinitrophenyl)-2-isopropylhydrazine; in lines 66 and 67, "**25**" should read 1-(2,4-dinitrophenyl)-2-cyclohexylhydrazine; in line 76, "Preparation of **22** from **25**" should read Preparation of 2-Isopropyl-6-nitrobenzotriazole 1-Oxide from 1-(2,4-Dinitrophenyl)-2-isopropylhydrazine."

Page 2983. Column 1. In line 1, "Synthesis of **23** from **25**" should read "Synthesis of 2-Cyclohexyl-6-nitrobenzotriazole 1-Oxide from 1-(2,4-Dinitrophenyl)-2-cyclohexylhydrazine."

David J. Pokorny and William W. Paudler*: Naphthyridine Chemistry. XIV. The Meisenheimer Reaction of the 1,X-Naphthyridine 1-Oxides.

Page 3105. Column 1, lines 15 and 63. Reference to E. M. Hawes and D. G. Wibberley [*J. Chem. Soc. C*, 1564 (1967)] reporting their syntheses and pmr spectra of 2-chloro- and 2-methoxy-1,8-naphthyridines was inadvertently left out. We wish to thank Dr. Hawes for bringing this to our attention.

R. A. Abramovitch,* G. Grins, R. B. Rogers, J. L. Atwood, M. D. Williams, and S. Crider: A Novel β -Alkylation of Pyridine and Quinoline 1-Oxides.

Page 3383. Column 1. In lines 23 and 24, " $J_{5,6} = 3$ Hz" should read $J_{5,6} = 6$ Hz; in lines 24 and 25, " $J_{4,5} = 4$, $J_{5,6} = 3$ Hz" should read $J_{4,5} = 8.9$, $J_{5,6} = 6$ Hz.

Louis A. Carpino* and Grace Y. Han: The 9-Fluorenylmethoxycarbonyl Amino-Protecting Group.

Page 3404. We have been informed of difficulties in the synthesis of 9-fluorenylmethanol. For most of the work described we applied the reproducible, although tedious, formylation of fluorene by means of potassium ethoxide in ether (86%) followed by cross-Cannizzaro reduction of the 9-formylfluorene (65%) [W. Wislicenus and M. Waldmuller, *Ber.*, **42**, 785 (1909); W. G. Brown and B. A. Bluestein, *J. Amer. Chem. Soc.*, **65**, 1082 (1943)]. More recently we have greatly simplified the method for the synthesis of this key alcohol by substituting sodium hydride for potassium metal and dispensing with the isolation of the intermediate aldehyde. The overall yield from fluorene is about the same as in the two-step procedure. The yield may be further increased by using a larger excess of both sodium hydride and

ethyl formate. On a large scale it is advantageous to recover unreacted fluorene by evaporation of the original ether extract.

9-Fluorenylmethanol.—A mixture of 8.3 g of fluorene, 8.5 ml of dry ethyl formate, 100 ml of anhydrous ether, and 6.5 g of NaH (57% dispersion in mineral oil) was refluxed with stirring for 3 hr and the resulting slurry was poured onto a mixture of ice and water such that the final aqueous volume was ~ 200 ml. An additional 75 ml of ether was added, the mixture was shaken in a separatory funnel, and the ether layer separated and was discarded. The aqueous layer was extracted once with 75 ml of ligroin (bp 60–70°), the ligroin was discarded, and the aqueous layer was stirred at room temperature for 1 hr after the addition of 22 ml of 35–40% aqueous formaldehyde solution. The off-white solid was filtered, washed twice with water, dried in air, and recrystallized from ligroin (bp 88–98°) to give 5.25 g (54.3%) of the alcohol as tiny papery needles: mp 100–101°; nmr ($CDCl_3$) δ 2.0 (s, 1, OH), 3.8–4.0 (m, 3, $CHCH_2$), 7.1–7.8 (m, 8, aryl). On the 0.05 *M* scale given above the yield can be raised to 60% by the use of 15 ml of ethyl formate and 10 g of sodium hydride dispersion.

R. K. Hill,* R. Soman, and S. Sawada: Asymmetric Induction in the Thermal Reactions of Allylic Alcohols with *N,N*-Dimethylacetamide Dimethyl Acetal and Triethyl Orthoacetate.

Page 3738. Column 2, line 6. The rotation of amide **V** is +15.7°, not –15.7°. Column 2, lines 23, 24, and 32. The configurations are *S*, not *R*. We thank Professor R. M. Magid, University of Tennessee, for calling our attention to these errors.

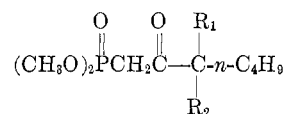
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R. A. Abramovitch* and B. W. Cue, Jr.: *N*-Hydroxypyrroles and Related Compounds.

Page 173. Column 1. In the summary, in line 3, "2-cyanopyrroles" should read *N*-hydroxy-2-cyanopyrroles. In line 4, "1-hydroxy-2-pyrrolones" should read 3-substituted 2,3-dihydro-2-pyrrolones.

M. Hayashi,* H. Miyake, T. Tanouchi, S. Iguchi, Y. Iguchi, and F. Tanouchi: The Synthesis of 16(*R*)- or 16(*S*)-Methylprostaglandins.

Page 1250. Column 2. Structure **2a**, **2b** should be



"**4c**" should read **4a**. For **2b**–**13b**, "16(*R*)-methyl series" should read 16(*S*)-methyl series.

Page 1251. Column 1, line 19. "**7a**" should read **7b**.

B. Goričnik, Z. Majerski, S. Borčić,* and D. E. Sunko: Secondary Deuterium Isotope Effects in the Solvolysis of Cyclobutyl and Cyclopropylcarbonyl Methanesulfonates.

Page 1885. Column 2, Acknowledgment. The acknowledgment is incomplete and should read as follows. This work has been supported in part by a grant from the Research Council of the Republic of Croatia and in part by PL 480 Grant administered by the National Institutes of Health, Bethesda, Md., Agreement No. 02-001-1.